

# A03: 「スピントロニクス 材料の探索」

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# 研究組織 「スピネレクトロニクス材料の探索」

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理論、デザイン

実験グループとの緊密な連携。

実証実験

スピネレクトロニクス材料のデザインとその実証。

理論、デザイン

# 役割分担

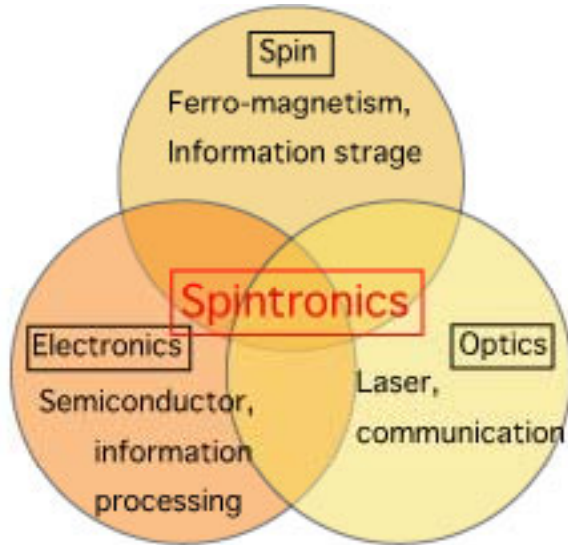
半導体スピントロニクス	デザイン	佐藤、下司、吉田	ナノ構造制御による半導体スピントロニクス
		小倉、赤井	N 半導体スピントロニクスデザイン
	実証実験	黒田、朝日	ナノ構造制御による半導体スピントロニクスデザイン 実証実験 ZnTe系、GaN系
金属系スピントロニクス	デザイン	小田	御 による 制
	実証実験	野崎、鈴木	実証実験
	実証実験		

LiZnAsベース磁性半導体の材料設計  
Computational materials design for LiZnAs-  
based magnetic semiconductors

藤、佐藤和則、吉田博  
阪大基礎工

S. Fujimoto, K. Sato, H. Katayama-Yoshida  
Osaka Univ.

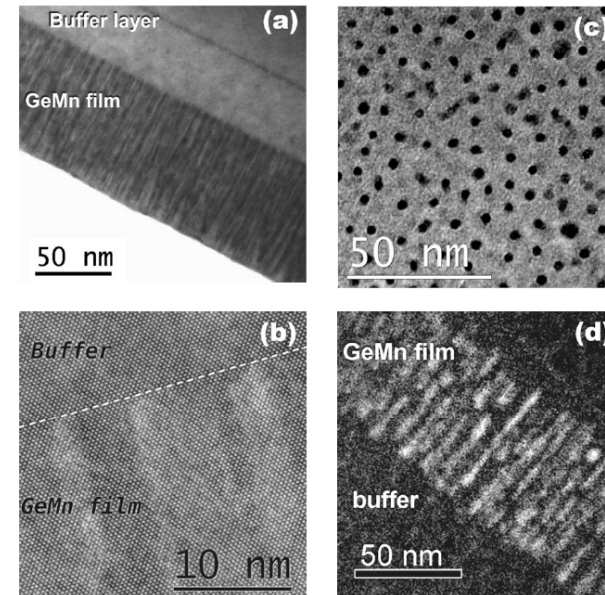
# Introduction



- ⌘ **Spintronics** = charge + spin
- ⌘ Dilute magnetic semiconductors(DMS)
  - Carrier induced ferromagnetism
  - (Ga, Mn)As, (In, Mn)As:  $T_C < RT$ .
- ⌘ **For high- $T_C$ , high concentration doping is necessary.**

S. A. Wolf et al, Science 294 (2001) 1488  
T. Dietl Semicond. Sci. Tech. 17 (2002) 377,  
K. Sato et al., Semicond. Sci. Tech. 17 (2002) 367

Homogeneous high concentration doping is difficult because of the **phase separation.**



Experiments. Singh et al., APL 86 (2005)12504  
Gu et al., JMMM 290-291(2005)1395.  
T. Devillers et al., PRB 76 (2007) 205306.

**Propose new materials, where homogeneous high concentration doping is possible. LiZnAs-based DMS**

# Filled tetrahedral semiconductors

Masek et al., PRL 98 (2007) 067202

## LiZnAs

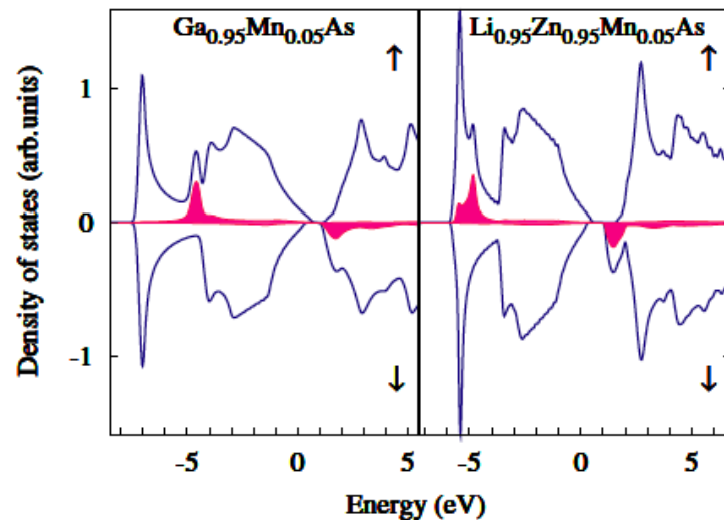
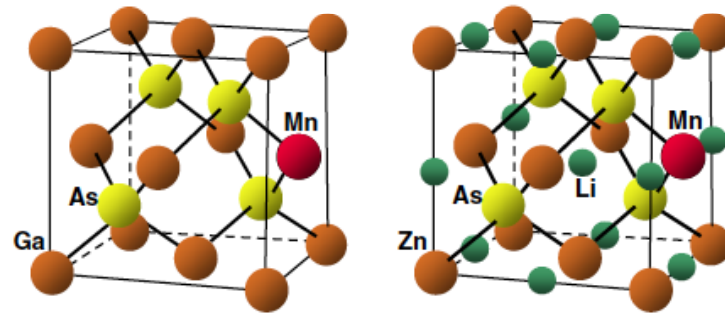


FIG. 1 (color online). Top panel: Schematics of (Ga,Mn)As and Li(Zn,Mn)As crystal structures. Bottom panel: *ab initio* total and Mn *d*-orbital projected density of states of  $\text{Ga}_{0.95}\text{Mn}_{0.05}\text{As}$  and  $\text{Li}_{0.95}(\text{Zn}_{0.95}, \text{Mn}_{0.05})\text{As}$  mixed crystals. Energy is measured from the Fermi energy.

- Zinc blende structure + interstitial Li (half-Heusler)
- $\text{Ga}^{3+}\text{As}^{3-}$   
 $\text{Zn}^{2+}\text{As}^{3-} + \text{Li}^{1+}$
- $\text{Zn}^{2+}$   $\text{Mn}^{2+}$ : High solubility
  - $E_{\text{Form}}(\text{Mn}_{\text{Zn}} \text{ in LiZnAs}) \approx -3$  (eV)
  - $E_{\text{Form}}(\text{Mn}_{\text{Ga}} \text{ in GaAs}) \approx -0.5$  (eV) (Mahadevan et al., PRB 68 (2003) 075202).
- Carrier control by Li addition (removal)
- Phase separation is also controlled by carrier doping.

Juza et al., Angew. Chem. 7 (1968) 360

v. G. Achenbach et al., Z. Anorg. Allg. Chem. 476 (1981) 9.

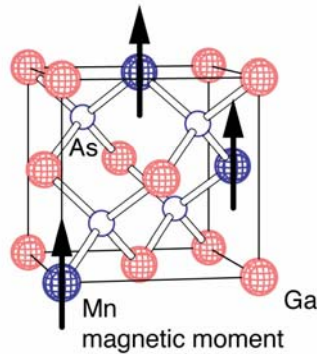
# Calculation of magnetic properties of DMS

K. Sato et al., RMP 82 (2010) 1633.

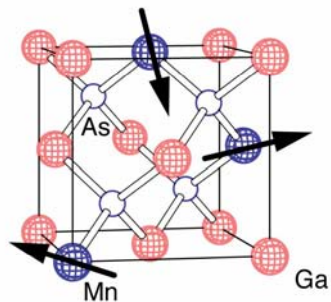
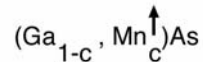
L. Begqvist et al., PRL 93 (2004) 137202

K. Sato et al., PRB 70 (2004) 201202

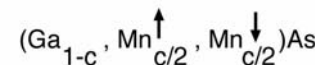
(Ga, Mn)As DMS



Ferromagnetic state

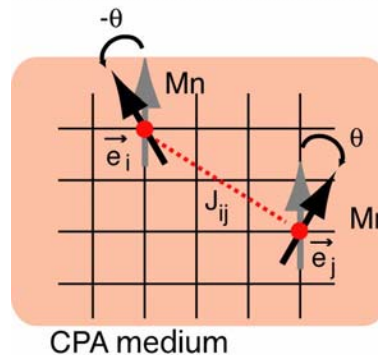


Paramagnetic state



⌘ Exchange interactions by Liechtenstein's formula

Energy difference due to the rotation is mapped to classical Heisenberg model (Liechtenstein et al.)



$$H = - \sum_{i \neq j} J_{ij} \vec{e}_i \cdot \vec{e}_j$$

$J_{ij}$  : Exchange interactions between 2 impurities in a CPA medium

$\vec{e}_i$  : direction of magnetic moment of the impurity at site i

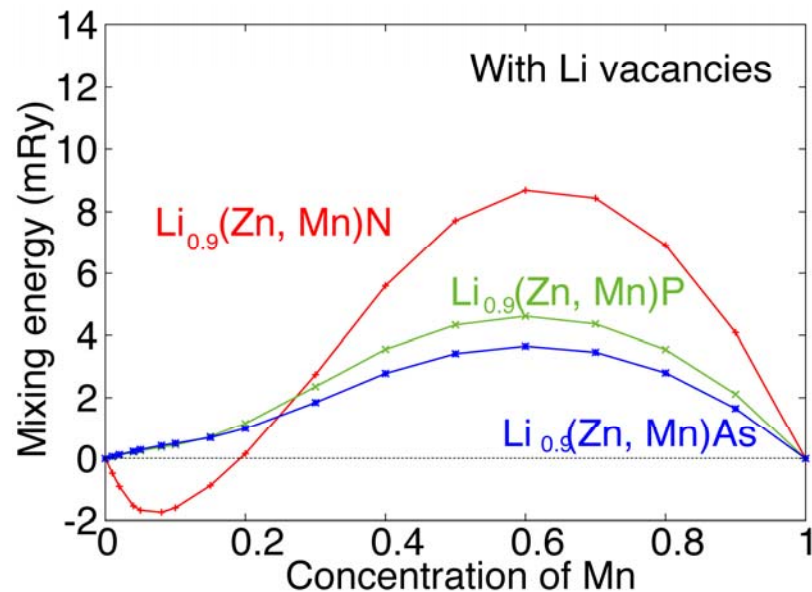
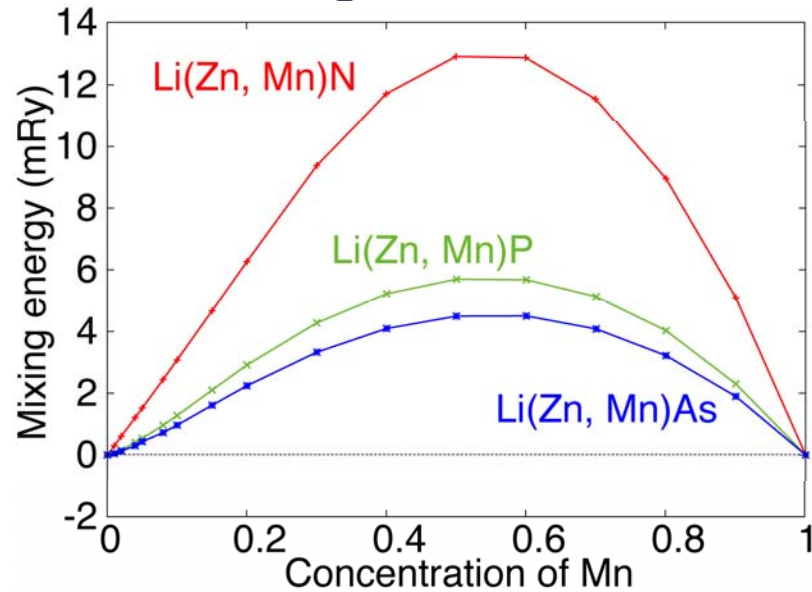
⌘ KKR-CPA-LDA

→ MACHIKANEYAMA2002 (Akai)

• Statistical method for  $T_C$

- Mean field approximation (MFA)
- Random phase approximation (RPA)
- Monte Carlo simulation (MCS)

# Stability of homogeneous Li(Zn, Mn)As



- Mixing energy of Li(Zn, Mn)As  

$$E_M = E[\text{LiZn}_{1-x}\text{Mn}_x\text{As}] - xE[\text{LiMnAs}] - (1-x)E[\text{LiZnAs}]$$

$E$ : Total energy

$x$ : concentration of Mn

- Upward convexity  
Phase separation
- Downward convexity  
homogeneous mixing  
ordered phase
- The presently investigated systems favor phase separation.

- By introducing  $V_{\text{Li}}$ , we can reduce the mixing energy.

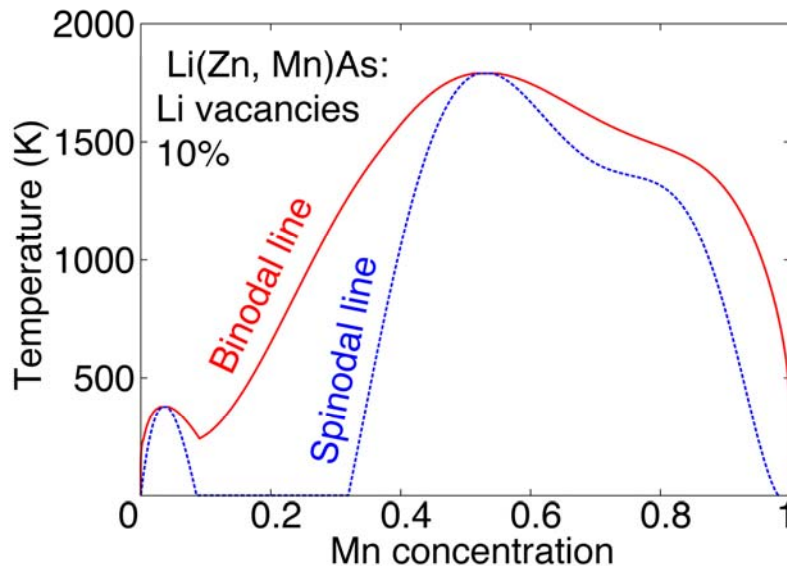
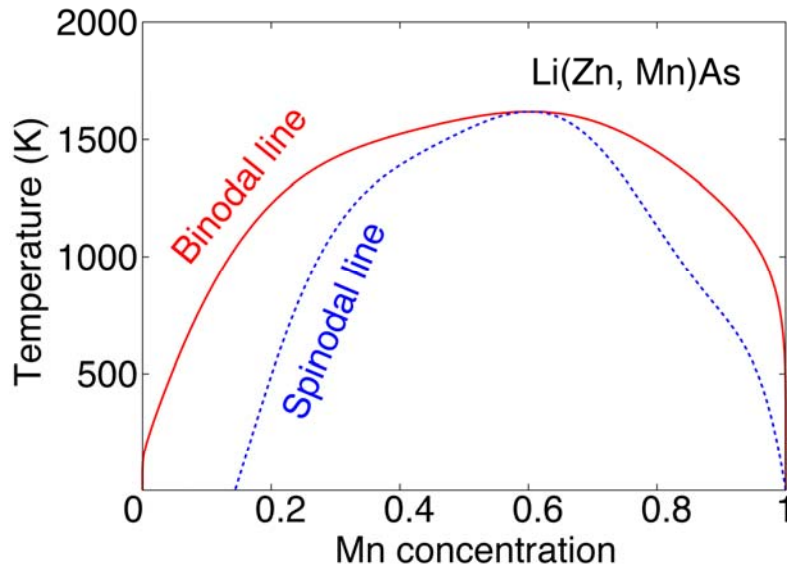
co-doping method:

L. Bergqvist et al. PRB 83 (2011) 165201.

H. Fujii et al., APEX 4 (2011) 043003.



# Phase diagram of Li(Zn, Mn)As



- Free energy  $F$  of two component alloy

$$F = E_M - TS$$

$T$ : temperature

$S$ : entropy

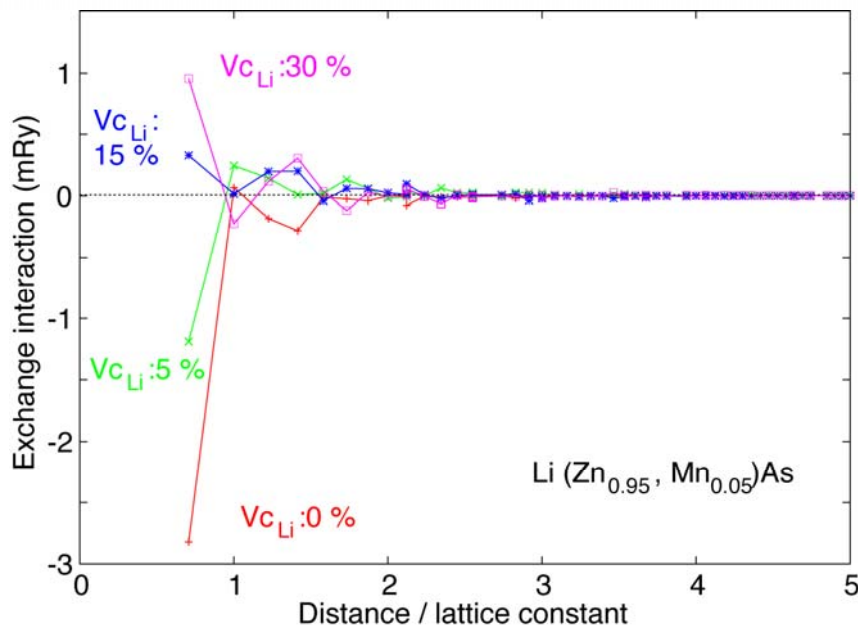
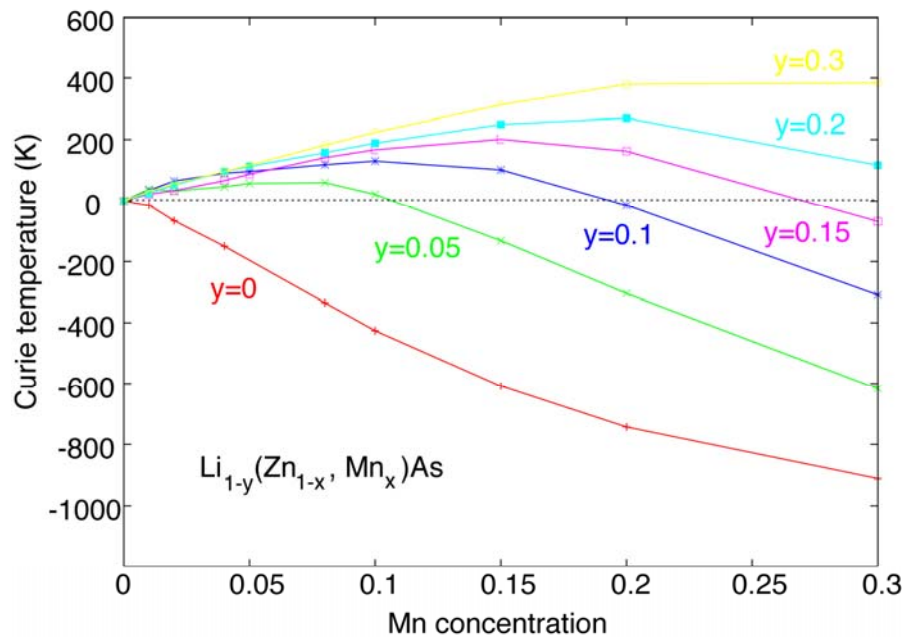
$$S = -k_B [(1-x) \log(1-x) + x \log(x)]$$

- Phase diagram

- Binodal line: common tangent

- Spinodal line:  $\frac{\partial^2 F}{\partial c^2} = 0$

- Spinodal decomposition is suppressed by introducing Li vacancies.



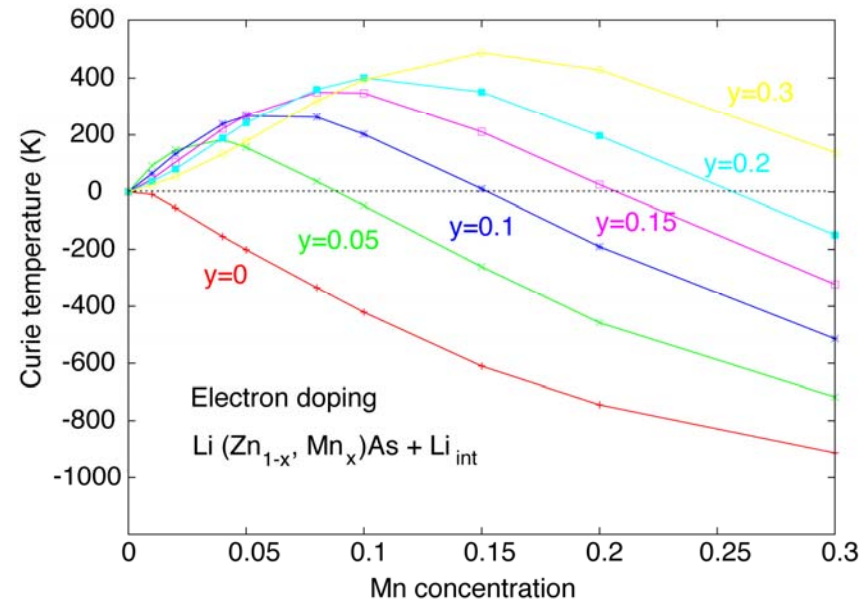
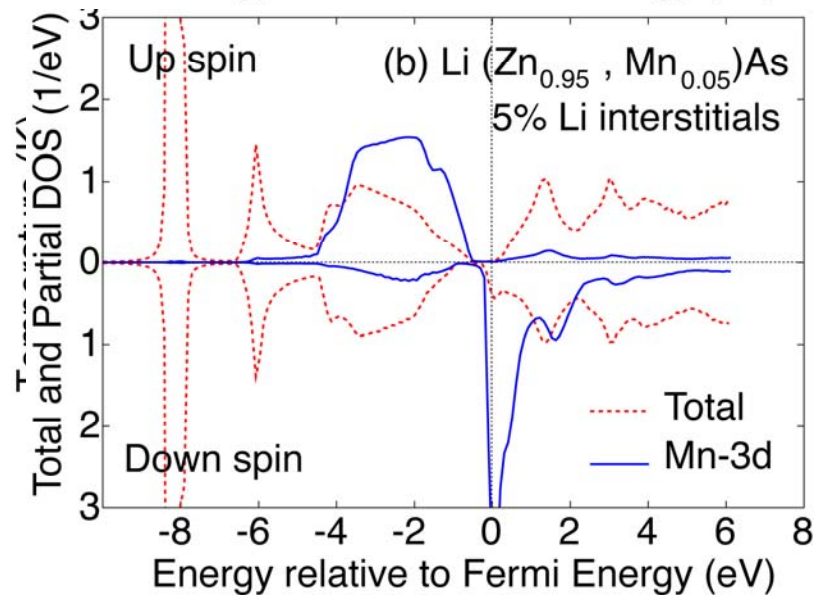
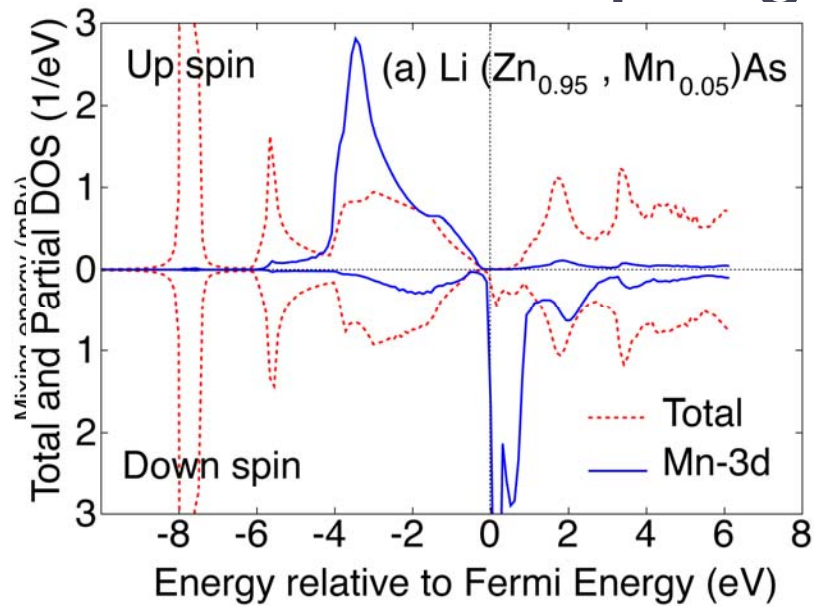
# $T_C$ (MFA) and DOS of $\text{Li}(\text{Zn}, \text{Mn})\text{As}$

- $T_C$  by the mean field approximation. PM: paramag. FM: ferromag.

$$k_B T_C = \frac{2}{3} \frac{E(\text{PM}) - E(\text{FM})}{x}$$

- $\text{Li}(\text{Zn}, \text{Mn})\text{As}$ 
  - Li vacancy introduces holes and induces ferromagnetism
  - $p$ - $d$  exchange mediates ferromagnetic interactions.

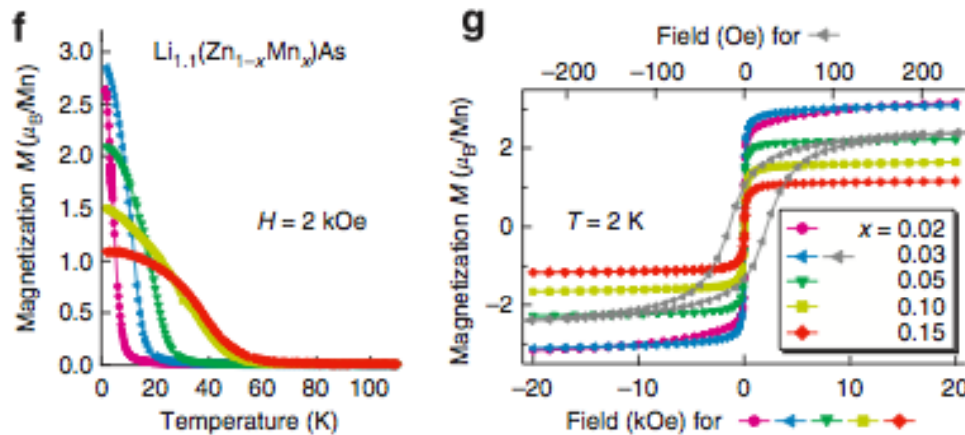
# Electron doping in Li(Zn, Mn)As



- Electron doping by **Li<sub>int</sub>**
  - $E_{\text{form}}(\text{Li}_{\text{int}} \text{ in LiZnAs}) = 0 \sim -1 \text{ (eV)}$
- Suppression of spinodal decomposition
- Ferromagnetism is induced by the electron doping.

# Recent experiment

Z. Deng et al., Nature Commun. 2:422, (2011, Aug.) 1-5.



- Solid-state reaction
- XRD
- dc-magnetization,  $\mu\text{SR}$
- Hall effect **p-type**
  - Li at Zn substitutional site

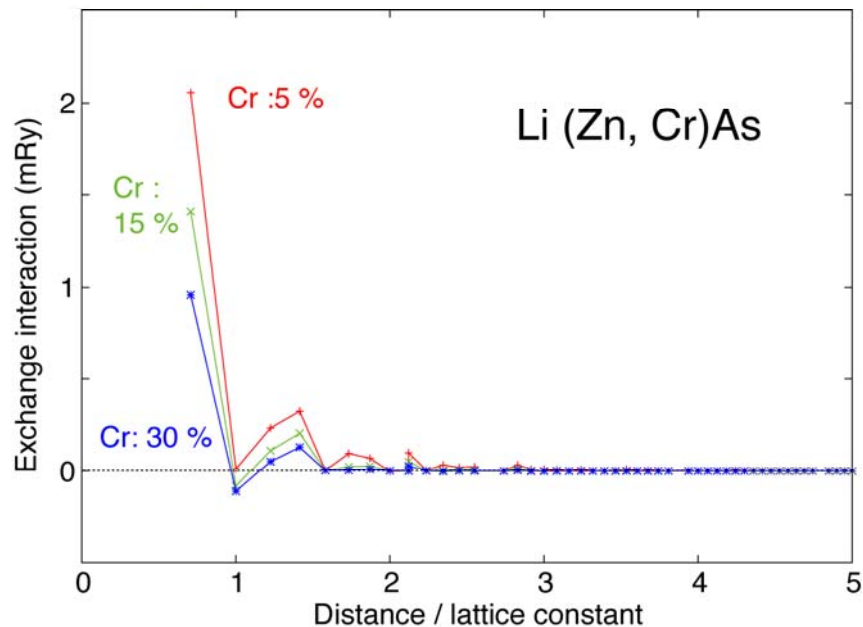
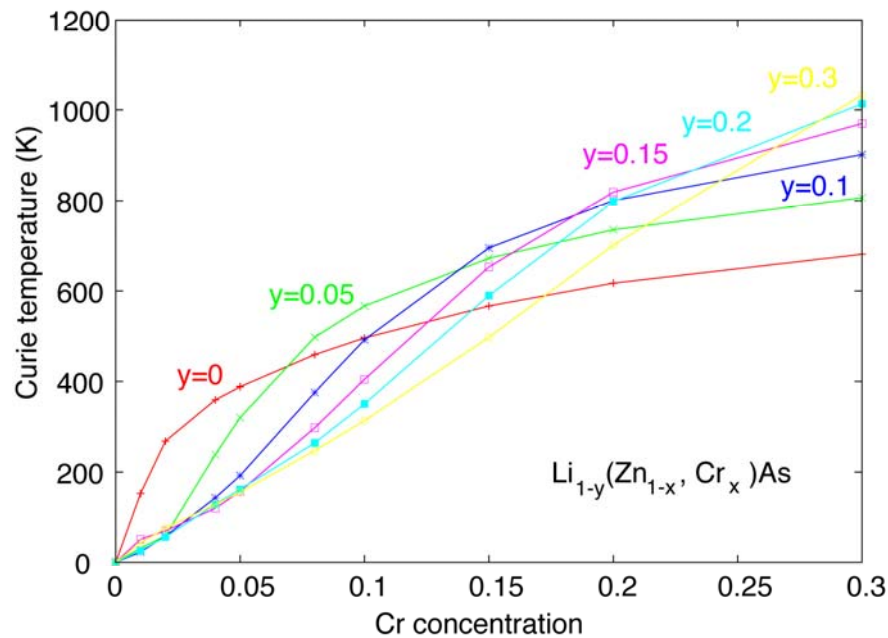
**Table 1 | Ferromagnetic transition temperature and ordered moment size.**

Li concentration	1.05		1.1		1.2	
	$T_c$ (K)	$M$ (T=2K)	$T_c$ (K)	$M$ (T=2K)	$T_c$ (K)	$M$ (T=2K)
<i>Mn concentration</i>						
0.03			17	2.9		
0.05			29	2.1	14	1.8
0.1	22	0.6	49	1.5	22	1.4
0.15	38	0.5	50	1.1	23	0.7

$M$  (Bohr magneton per Mn).

The values of  $T_c$  and the average ordered moment size  $M$  per Mn of  $\text{Li}_{1-x}(\text{Zn}_{1-x}\text{Mn}_x)\text{As}$  at  $T=2\text{K}$  and  $H=2\text{kOe}$  derived from magnetization measurements for several combinations of Li and Mn concentrations.

# $T_C$ (MFA) and DOS of Li(Zn, Cr)As



- $T_C$  is calculated by the mean field approximation.
- Li(Zn, Cr)As
  - Ferromagnetic state is stable.
  - double exchange mechanism dominates
  - Hole doping is effective to raise  $T_C$  for Cr high concentration region.

# Summary

- First-principles  $T_C$  calc. of DMS
  - Hybrid method = ab initio exchange const. + Monte Carlo
  - Reliable  $T_C$  calculations are possible.
  - For high- $T_C$ , high concentration doping is needed, however solubility of Mn is very low in DMS.
- New material: Li(Zn, Mn)As
  1. Formation energy of  $\text{Mn}_{\text{Zn}}$  is negative.
  2. By introducing  $V_{\text{Li}}$ ,  $\text{Li}_{\text{int}}$  ('co-doping'), we can control the phase separation and magnetism.
  3. Recent experiments:  $\text{Li}_{1.1}(\text{Zn}_{0.85}, \text{Mn}_{0.15})\text{As}$ ,  $T_C \sim 50$  (K)
  4. Cr doping is also promising.